First-Principles Calculations of the Structural, Electronic and Magnetic Properties of Mn-Doped InSb by Using mBJ Approximation for Spintronic Application

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1. Introduction

In September 1925, Goudsmith and Uhlenbeck discovered the spin [1]. This discovery was a radical change for semiconductors, but some years later the magnetic structure for electrons led to spintronics. Spintronics or magnetoelectronics is important for applications such as to store information, nuclear magnetic resonance (NMR), electron spin resonance spectroscopy, and magnetic resonance imaging (MRI) in medicine. The real importance of spintronics was the giant magnetoresistance (GMR) discovered by Barnaś and Fert in 1988 [2] for drive head technology which are used in modern hard disks. The dilute magnetic semiconductors (DMSs) are semiconductors which can be doped with transition metals or rare earth elements, leading to a variety of cooperative effects. In the early 1970s and late 1980s a new diluted magnetic III–V semiconductor In$_{1-x}$Mn$_x$As (with a low concentration) has been produced by molecular-beam epitaxy (MBE) [3]. Recently, many researchers are working on diluted magnetic III–V and II–VI semiconductors for spintronics applications. Berber et al. have predicted the half-metallic ferromagnetism for Sr$_{1-x}$V$_x$O and found that vanadium (V) doped strontium oxide (SrO) is a half-metallic ferromagnet for a concentration $x = 0.25$ and 0.50 with a half-metallic ferromagnetic gap (G$_{HM}$) of 2.634, 2.359 eV and a half-metallic gap (G$_{HM}$) of 1.0, 0.1 eV respectively [4, 5]. Berriah et al. studied half-metallicity of Cr doped BaTe and predicted that for all concentrations of Cr the structure Ba$_{1-x}$Cr$_x$Te is a half-metallic ferromagnet with a spin polarization of 100% [6]. The considerable $sp - d$ exchange interaction between the electrons of the band and Mn$^{2+}$, the DMSs II–VI and III–V doped with manganese (Mn) attracts many researchers. The prediction that was made [7] of Mn-doped AlSb (Al$_{1-x}$Mn$_x$Sb) is a half-metal ferromagnet for all concentrations (except for MnSb). The magnetic, magnetotransport and magnetooptical properties were studied by Ohno for Mn-doped GaAs and Mn-doped InAs [8].

In this context we have chosen to study the electronic and magnetic properties of the binary InSb doped with manganese (Mn). In this study, we have predicted the structural, electronic and magnetic properties of In$_{1-x}$Mn$_x$Sb alloys with the concentrations ($x$) ($x = 0.25, 0.50$ and 0.75). For this prediction, we have utilized the full potential linearized augmented plane wave (FP–LAPW) method within density functional theory (DFT) [9]. For theoretical calculations of the structural constants we used the generalized gradient approximation GGA with the WC functional [10], but for the electronic and magnetic properties we used the Tran–Blaha modified Becke–Johnson exchange potential approximation (TB-mBJ) [11].

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This paper is composed as follows: in Sect. 2, we describe the method of the calculation. In Sect. 3, we present the details of the collected results and discuss the structural, electronic, and magnetic properties of the In$_{1-x}$Mn$_x$Sb alloys. The main conclusions of our present work are summarized in Sect. 4.

2. Computational detail

In this paper, we have used the full-potential linearized augmented plane wave (FP-LAPW) method within the framework of the density functional theory (DFT) [9] as implemented in the WIEN2k code [12]. For computing the structural parameters, we have used the generalized gradient approximation (GGA) with the WC functional [10]. The electronic and magnetic properties of In$_{1-x}$Mn$_x$Sb are calculated using (GGA-WC), but with the new favored and improved method designed, known as the TB-mBJ potential approach [11] which enables us to perfectly represent the electronic structure of solids [13–15]. For the evaluation of the band gap, this theory has been proven by Tran [16] to be better than the GGA and LDA exchange correlation potential. In the present investigation, we have the muffin-tin radii (MT) for all atoms (In, Mn, and Sb) to be 2.50 atomic units (a.u.). The $K_{\text{max}} = 9(R_{\text{MT}})^{-1}$ ($K_{\text{max}}$ is the plane wave cut-off and $R_{\text{MT}}$ is the smallest of all atomic sphere radii). The Fourier expansion of the charge density was truncated at $G_{\text{max}} = 12$ (Ry)$^{1/2}$, the $l$-expansion of the non-spherical potential and charge density was carried out up to $l_{\text{max}} = 10$. The cut-off energy is set to −7 Ry to separate the core from the valence states. The self-consistent calculations are evaluated to be converged when the total energy of the system is stable within 0.0001 Ry.

3. Results and discussion

3.1. Stability and structural properties

Indium antimonide (InSb) crystallizes in the zinc-blende structure (no. 216), in which the two types of atoms, In and Sb, are located at (0, 0, 0) and (0.25, 0.25, 0.25) positions, respectively. We created a supercells with 8 atoms (4 atoms for In and 4 for Sb). We substituted one or three of these sites to create the structure of In$_{0.75}$Mn$_{0.25}$Sb and In$_{0.25}$Mn$_{0.75}$Sb with concentration $x = 0.25$, and 0.75, respectively. Then we created a supercells with 16 atoms (8 atoms for In and 8 for Sb), and substituted In sites with Mn to create the structure In$_{0.50}$Mn$_{0.50}$Sb. The In$_{0.75}$Mn$_{0.25}$Sb and In$_{0.25}$Mn$_{0.75}$Sb structures have a cubic structure with space group P-43m (no. 215), while In$_{0.50}$Mn$_{0.50}$Sb has a cubic structure in space group F-43m (no. 216) (zinc blende).

<table>
<thead>
<tr>
<th>Compound</th>
<th>$a$ [Å]</th>
<th>$B$</th>
<th>$B'$</th>
<th>Method</th>
</tr>
</thead>
<tbody>
<tr>
<td>InSb</td>
<td>6.5555, 6.464°, 6.479°, 6.629°</td>
<td>42.4612, 37.368°</td>
<td>4.0525, 4.452°</td>
<td>GGA-WC</td>
</tr>
<tr>
<td>In$<em>{0.75}$Mn$</em>{0.25}$Sb</td>
<td>6.4363</td>
<td>44.3600</td>
<td>6.0030</td>
<td>GGA-WC</td>
</tr>
<tr>
<td>In$<em>{0.50}$Mn$</em>{0.50}$Sb</td>
<td>6.3339</td>
<td>48.0342</td>
<td>4.2197</td>
<td>GGA-WC</td>
</tr>
<tr>
<td>In$<em>{0.25}$Mn$</em>{0.75}$Sb</td>
<td>6.2021</td>
<td>44.0995</td>
<td>6.8125</td>
<td>GGA-WC</td>
</tr>
</tbody>
</table>

After optimization of all structures of In$_{1-x}$Mn$_x$Sb with $x = 0.25, 0.50, 0.75, 1$, we calculated the total energies as a functional the unit cell volumes and fitted the results with the Birch–Murnaghan equation of state (EOS) [17] to evaluate the ground state properties such as the equilibrium lattice constant $a$, the bulk modulus $B_0$ and the bulk modulus pressure derivative $B'$.

The variation of lattice constant $a$ as a function of the concentration ($x$) for In$_{1-x}$Mn$_x$Sb is illustrated in Fig. 1. This variation of the calculated lattice parameter versus concentration $x$ is fitted to a polynomial of second order $a(x) = 6.55358 - 0.43389x - 0.04011x^2$. (1) The thermodynamic stability for the synthesis of the alloys is determined by the energy of formation [21, 22]. In order to obtain the phase stability of the In$_{0.75}$Mn$_{0.25}$Sb, In$_{0.50}$Mn$_{0.50}$Sb, In$_{0.25}$Mn$_{0.75}$Sb structures, we computed the formation energies using the following formula [23, 24]:

$$E_f = E_{\text{total}}^{\text{In}_{1-x}\text{Mn}_x\text{Sb}} - (xE_{\text{Sb}}^\text{Bulk} + xE_{\text{Mn}}^\text{Bulk} + (1 - x)E_{\text{In}}^\text{Bulk}),$$  

where $E_{\text{total}}^{\text{In}_{1-x}\text{Mn}_x\text{Sb}}$ is the total energy per atom of In$_{1-x}$Mn$_x$Sb, $x$ is the concentration of Mn in the supercell. The $E_{\text{Sb}}^\text{Bulk}$, $E_{\text{Mn}}^\text{Bulk}$ and $E_{\text{In}}^\text{Bulk}$ are the total energies of bulk Sb, Mn, and In, respectively. Our calculated values of the formation energy for the ternary alloys are $-2.35,$
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Fig. 1. Calculated lattice constant $a$ of In$_{1-x}$Mn$_x$Sb as a function of concentration $x$.

$-2.04$, and $-3.18$ eV for In$_{0.75}$Mn$_{0.25}$Sb, In$_{0.50}$Mn$_{0.50}$Sb, and In$_{0.25}$Mn$_{0.75}$Sb, respectively. Therefore, the negative values of formation energies signify that the ternary alloys In$_{1-x}$Mn$_x$Sb ($x = 0.25$, 0.50 and 0.75) are thermodynamically stable and can be synthesized.

3.2. Electronic, magnetic properties and half-metallic character

In this part, we present the electronic band structure, total and partial densities of states for In$_{1-x}$Mn$_x$Sb with different concentration $x$ ($x = 0$, 0.25, 0.50, 0.75, and 1) using the GGA-WC [10] of TB-mBJ [11] approaches. The band structures of the alloys of In$_{1-x}$Mn$_x$Sb are shown in Figs. 2–6 with the concentrations 0, 0.25, 0.50, 0.75, and 1, respectively. Since InSb is non-magnetic, the two spin channels of InSb have the same semiconducting band structures with a direct band gap ($E_{\Gamma - \Gamma}$) located at the $\Gamma$ point. The spin polarized band structures of In$_{1-x}$Mn$_x$Sb (at concentrations of $x = 0.25$, 0.50, 0.75, and 1) are presented in Figs. 3–6. We remark that the majority-spin bands are metallic whereas a gap is located at the Fermi level for the minority-spin bands. For the minority-spin, we notice two gaps, the half-metallic ferromagnetic gap ($G_{\text{HMF}}$) and the half-metallic gap ($G_{\text{HM}}$). The $G_{\text{HMF}}$ is the energy difference between the valence-band maximum (VBM) and conduction-band minimum (CBM), while $G_{\text{HM}}$ is the (smaller) energy between the Fermi level and the maximum value of the valence band energy [25, 26].

Table II illustrates the computed direct gap ($E_{\Gamma - \Gamma}$) of InSb, and the HMF ($G_{\text{HMF}}$) and HM ($G_{\text{HM}}$) gaps of In$_{1-x}$Mn$_x$Sb at all compositions ($x$), using the TB-mBJ exchange potential, together with other experimental and theoretical data [20]. Figures 3 and 4 show that In$_{0.75}$Mn$_{0.25}$Sb and In$_{0.50}$Mn$_{0.50}$Sb have a direct half-metallic ferromagnetic gap $G_{\text{HMF}}$ of 1.4085 and 1.3192 eV at the $\Gamma$ high-symmetry point and exhibit a half-metallic gap $G_{\text{HM}}$ of 0.3161 and 0.4769 eV, respectively. However, Fig. 5 illustrates that In$_{0.25}$Mn$_{0.75}$Sb shows an indirect half-metallic ferromagnetic gap between $R$ and $\Gamma$ high-symmetry points of 1.4503 eV and a half-metallic gap of
0.4140 eV. This makes them promising materials for spintronic applications. In summary, Table II shows that our results have improved the results of Verma et al. [20].

The densities of states (DOS) near the Fermi level $E_F$ describe the metallic or semiconductor behaviour of these structures, based on the ferromagnetic disposition; we have also calculated the spin-polarized total DOS (TDOS) and partial densities of states (PDOS) for In$_{1-x}$Mn$_x$Sb for different compositions ($x$). The spin-polarized (TDOS) and (PDOS) of InSb, In$_{0.75}$Mn$_{0.25}$Sb, In$_{0.50}$Mn$_{0.50}$Sb, and In$_{0.25}$Mn$_{0.75}$Sb are demonstrated in Figs. 6–9, respectively. For InSb, the spin-up and spin-down states are the same with a semiconductor character, and this non-magnetic type of this compound is exposed in Fig. 6. On the other hand, Figs. 7–9 exhibit the TDOS of In$_{1-x}$Mn$_x$Sb for all concentrations with different band states for spin-up and spin-down, caused by the different $p$–$d$ hybridization between the Sb $p$ and Mn 3$d$ states around the Fermi level for the two directions. The minority-spin states (for all compositions) show a gap around the Fermi level, and thus In$_{0.75}$Mn$_{0.25}$Sb, In$_{0.50}$Mn$_{0.50}$Sb, and In$_{0.25}$Mn$_{0.75}$Sb alloys are half-metallic ferromagnets with a spin polarization of 100% according to the following expression [27]:

$$ P = \frac{N \uparrow (E_F) - N \downarrow (E_F)}{N \uparrow (E_F) + N \downarrow (E_F)} \times 100\%. $$

(3)

$P$ is the spin polarization, where $N \uparrow (E_F)$ and $N \downarrow (E_F)$ are the densities of states of the majority and minority spin around the Fermi level, respectively. We observe
TABLE II
Calculated half-metallic ferromagnetic gap ($G_{\text{HMF}}$) and half-metallic gap ($G_{\text{HM}}$) of minority-spin band for In$_{1-x}$Mn$_x$Sb with composition $x = 0.25$, 0.5, and 0.75 and direct band gap $E_g$ for InSb.

<table>
<thead>
<tr>
<th>Compound</th>
<th>$G_{\text{HMF}}$ [eV]</th>
<th>$G_{\text{HM}}$ [eV]</th>
<th>$E_g$ [eV]</th>
<th>Method</th>
<th>Behavior</th>
</tr>
</thead>
<tbody>
<tr>
<td>InSb</td>
<td>–</td>
<td>–</td>
<td>0.3216, 0.24$^a$</td>
<td>TB-mBJ</td>
<td>HMF</td>
</tr>
<tr>
<td>In$<em>{0.75}$Mn$</em>{0.25}$Sb</td>
<td>1.4085, 0.328$^c$</td>
<td>0.3161, 0.045$^c$</td>
<td></td>
<td>TB-mBJ</td>
<td>HMF</td>
</tr>
<tr>
<td>In$<em>{0.50}$Mn$</em>{0.50}$Sb</td>
<td>1.3192, 0.885$^c$</td>
<td>0.4709, 0.395$^c$</td>
<td></td>
<td>TB-mBJ</td>
<td>HMF</td>
</tr>
<tr>
<td>In$<em>{0.25}$Mn$</em>{0.75}$Sb</td>
<td>1.4503, 0.961$^c$</td>
<td>0.4140, 0.250$^c$</td>
<td></td>
<td>TB-mBJ</td>
<td>HMF</td>
</tr>
</tbody>
</table>

$^a$ Ref. [19], $^c$ Ref. [20].

that the top section of the valence bands is primarily composed of Mn 3d partially occupied states, while the lowest part of the conduction bands is controlled by the Mn 3d unoccupied states, which are higher than the Fermi level $E_F$.

Finally, Table III exhibits the total and local magnetic moments of In$_{0.75}$Mn$_{0.25}$Sb, In$_{0.50}$Mn$_{0.50}$Sb, and In$_{0.25}$Mn$_{0.75}$Sb. We calculated the total and local magnetic moments of the relevant Mn, In, and Sb atoms, and at the interstitial region in the In$_{1-x}$Mn$_x$Sb structures in order to get the details of the $p - d$ exchange system on the magnetic properties. The results show that the total magnetic moment for all compounds was 4 $\mu_B$, dominated by the local magnetic moment of the Mn atom. The Mn 3d majority-spin states, which are partially filled with five electrons, cause a total magnetic moment of 4 $\mu_B$. On the other hand, the simply predicted magnetic moment of the Mn atom is reduced to below 4 $\mu_B$ and minor local magnetic moments are
induced at In, Sb, and the interstitial regions due to the $p - d$ exchange interaction between the Sb $p$ and Mn $3d$ levels. Table III shows that our results have improved the results of Verma et al. [20]. It also indicates that for all compositions of In$_{1-x}$Mn$_x$Sb, not only the magnetic moments of the Mn and Sb but also of Mn and In have opposite signs, indicating an antiferromagnetic interaction between Mn and Sb and between Mn and In.

4. Conclusion

In this study, we have used the first-principle calculations of the structural, electronic, and magnetic properties of the In$_{1-x}$Mn$_x$Sb alloys ($x = 0.25, 0.50$ and $0.75$). We have exploited the GGA-WC exchange correlation energy and improved it with the TB-mBJ approximation. The calculated formation energy indicates that the In$_{1-x}$Mn$_x$Sb compounds (for all section compositions) are thermodynamically stable, and can be synthesized. The computed lattice constant and energy band gap for InSb are in good agreement with the available experimental and theoretical data. The examination of the band structure and the density of states determine the half-metallic behaviour of In$_{1-x}$Mn$_x$Sb for all concentrations. The decomposition of the calculated total magnetic moment shows an antiferromagnetic interaction between Mn and Sb and between Mn and In. Therefore, the In$_{1-x}$Mn$_x$Sb compounds are suitable candidates for possible spintronics applications, and thus we hope that our results can serve as reference for theoretical and experimental studies in future.

### Table III

Calculated total and local magnetic moments (in the Bohr magneton $\mu_B$) within the muffin-tin spheres and in the interstitial sites for In$_{1-x}$Mn$_x$Sb with composition $x = 0.25, 0.50$, and $0.75$.

<table>
<thead>
<tr>
<th>Compound</th>
<th>Total [(\mu_B)]</th>
<th>In [(\mu_B)]</th>
<th>Mn [(\mu_B)]</th>
<th>Sb [(\mu_B)]</th>
<th>Interstitial [(\mu_B)]</th>
</tr>
</thead>
<tbody>
<tr>
<td>In$<em>{0.75}$Mn$</em>{0.25}$Sb</td>
<td>4, 4.00975$^c$</td>
<td>0.01781, 0.0049$^c$</td>
<td>3.86920, 3.8074$^c$</td>
<td>$-0.05002, -0.0087^c$</td>
<td>0.26823, 0.7994$^c$</td>
</tr>
<tr>
<td>In$<em>{0.50}$Mn$</em>{0.50}$Sb</td>
<td>4, 4.001475$^c$</td>
<td>0.02340, 0.0195$^c$</td>
<td>4.01154, 3.7579$^c$</td>
<td>$-0.35069, -0.0106^c$</td>
<td>0.76202, 1.6796$^c$</td>
</tr>
<tr>
<td>In$<em>{0.25}$Mn$</em>{0.75}$Sb</td>
<td>4, 4.00185$^c$</td>
<td>0.06886, 0.0405$^c$</td>
<td>3.94111, 3.7131$^c$</td>
<td>$-0.10514, -0.0768^c$</td>
<td>0.71483, 2.7009$^c$</td>
</tr>
</tbody>
</table>

$^c$ Ref. [20].

### References