

The NO₂ Sensing Properties of the Sensors Done with Nano-Tetrapods

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In this work we have studied the NO₂ sensing properties of the sensors with tetrapods. The tetrapods that were used in the production were obtained by CVD route, and their structural analysis was carried out using SEM, XRD and EDX methods. The sensors were produced by a different method than the one mentioned in our another work, and additional electrical $I - V$ characterization was carried out. As well as in the previous work the sensors were tested for 50 ppm sensitivity, by using computer controlled gas flux system, in NO₂ at 200 °C, and in ambient air to model the normal working conditions more closely. It was found that the sensors have a good sensitivity to the NO₂ gas. All of the sensors showed a great reversibility and it was found that the recovery time was perfectly low. After this experiment the same sensors were tested with other different gases for several times. And no other problems in sensing ability were seen.

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1. Introduction

Due to a direct and wide bandgap (3.37 eV at room temperature), ZnO is considered to be a very important semiconductor. Due to its various properties like wide band gap [1, 2], good thermal and chemical stability [3, 4], antimicrobial [5, 6], photocatalytic and sensor [7, 8] features, it is a promising material for potential future applications [9]. One of the expected application of ZnO is in the area of biocompatible gas sensors which can be used to detect gases like NO₂, hazardous for the nature and human health.

In the present work we have synthesized randomly precipitated and perfectly formed ZnO tetrapods in APCVD system in a constant oxygen flow at 1200 °C. The nanostructures were characterized from the aspect of nanostructure, XRD and EDX characteristics. It was found that the structures were perfect pure wurtzite crystal tetrapods. The length of the structures was measured to be approximately 10 to 18 μm and it was found that the thickness of the legs was between approximately 140 nm and 40 nm. The tetrapods were used to produce a gas sensor by trying a novel method. The obtained sensor has been able to perfectly detect presence of 50 ppm of NO₂ in the air at 200 °C.

2. Experimental

We started the experiments by preparing the Si substrates for the APCVD system. As the first step Si

substrates were cleaned in sonic vibrator in acetone (99.999%) for 10 min and in UP water for 10 min for 5 times. Then ZnO tetrapod-shaped nanostructures were grown on the substrates using an APCVD system. The nano-tetrapods were synthesized on Si (111) substrates in a horizontal tube furnace system, under the constant flow of oxygen and argon, used as the carrier gas. Pure ZnO (99.9%, 0.03 g) was placed inside the furnace system and the Si substrate was placed in the precipitation region of the system. To synthesize the ZnO, the zinc source material was vaporized and precipitated on the Si substrate at atmospheric pressure under flow of Ar (99.999%) at 120 sccm and O₂ (99.999%) at 5 sccm gas mixture. The temperature of the system was gradually increased to 1200 °C and then cooled down to room temperature.

As the third step the Si substrates for the sensors were prepared in a thermal evaporator system using Au (99.999%). The gold was deposited through a mask which allows to make the heating and conducting network for the sensors. The deposited film thickness was 20 nm. After the Au deposition, the samples were cleaned by a subsequent rinsing in acetone, ethanol, and water. As the fourth step, the obtained tetrapods were mixed with ethanol and applied on the surface of the SiO₂ using a pasteur pipette. Then the substrates were dried at 100 °C for 12 hours.

As fifth and the last step the sensitivity measurements were carried out in a computer controlled closed gas system, under 150 sccm of ambient dry air and 50 sccm of NO₂ gas at 200 °C. The gas sensing properties of the tetrapods were tested by measuring the resistance change due to the gas adsorbed on their surface.

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3. Results and discussion

The prepared ZnO tetrapods were examined in a SEM system and analyzed with EDX. We observed nanosized tetrapods had a morphology very similar to a symmetric tetrapod system. The fact that breaks the symmetry is the angle between the legs of the tetrapods, which is assumed to be caused of the randomness inside the flow system. The length of the structures were measured to be approximately 10 to 18 μm , as can be seen in SEM image (Fig. 1). The thickness of the legs of the tetrapods was found to be between 40 nm and 140 nm. To determine the content of the tetrapods, the nanoparticles were studied using EDX system. Figure 2 shows that EDX measurement contains only the characteristic peaks of Zn and O, which demonstrates that tetrapods do not contain impurities.

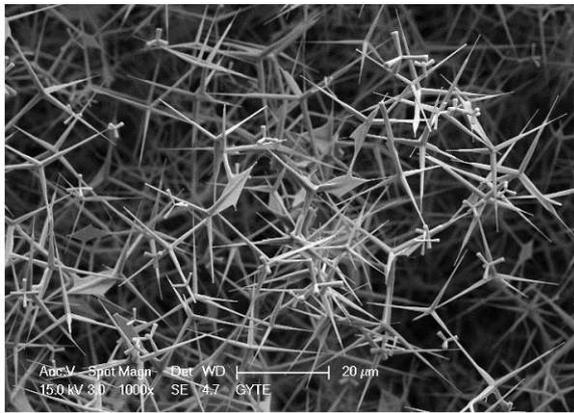


Fig. 1. SEM of the nano-tetrapods.

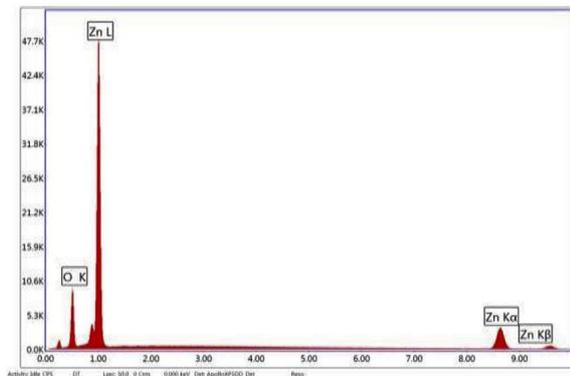


Fig. 2. EDX spectrum of the nano-tetrapods.

Figure 3 depicts the XRD diagram of the nano tetrapods. All diffraction peaks well match with the wurtzite hexagonal structure of ZnO, thus confirming that the material of the tetrapods is zinc oxide.

The sensors were characterized in terms of NO_2 sensing properties. The NO_2 gas was introduced into the gas chamber by using a syringe (50 sccm). Because of the good mobility of NO_2 and high temperature inside the closed gas cell, no additional actions were needed to obtain a homogeneous gas distribution.

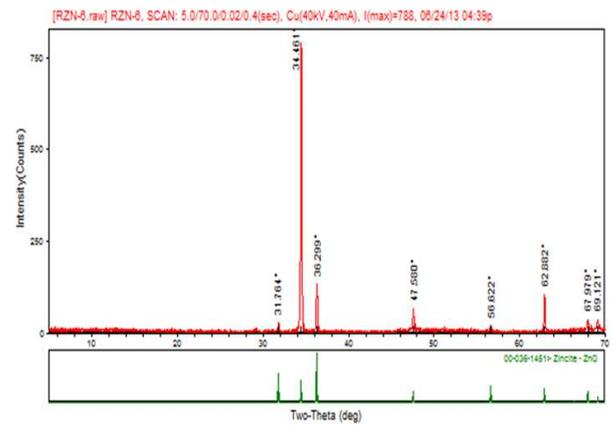


Fig. 3. XRD diagram of the nano-tetrapods used in the studied sensor.

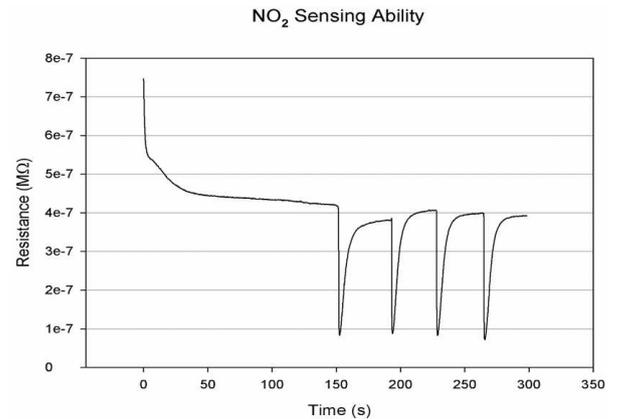


Fig. 4. NO_2 Gas sensing ability of the sensor with nano-tetrapods.

Undoped ZnO is an n-type semiconductor and due to the fact that NO_2 is an oxidizing gas, it accepts the electrons during a surface reaction [10]. This adsorption of NO_2 results in an increment of the width of the surface charge layer [11]. This phenomenon causes the sensor resistance to increase, as can be seen in Fig. 4. Each time the NO_2 gas was introduced into the system, the resistance had changed dramatically. Another point which is very important for the experiment, is that before and after the experiments no loss of nanowires' sensing ability was observed. The sensors were found do preserve all nanowires on their surface and the measurement could be taken nearly 50–60 times. During this entire period of time the information obtained from the sensors was correct and repeatable.

4. Conclusions

In the presented work we have produced a new sensor by using high-surface-area nano-tetrapods, using a five step method. The tetrapods were characterized in terms of their nanostructure, chemical composition, crystal structure and NO_2 sensing mechanism. The crucial

point of this work was making the sensor surface durable against the gas flow because of the low weight of the nanowires. We can say that the technique we have applied to nanowires to make them stay on the surface of the sensor worked well, and has allowed us to obtain meaningful data from the sensors, as can be seen from the measurement results.

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