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Dielectric and Conductivity Anisotropy in Liquid Crystalline Phases of Strongly Polar Thioesters

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Dedicated to Professor Jerzy Janik on the occasion of his 80th birthday

The main objective of this paper is to study dielectric properties of two compounds of acronyms 5OSCl and 6OSCl having strongly polar bond (C-Cl) at *para* position. Dielectric measurements were done in the frequency range from 40 Hz to 15 MHz. The dielectric spectra were measured for two principal alignments. For homeotropic alignment, the reorientation of molecule around the short molecular axis was observed. Both substances studied exhibit large positive dielectric anisotropy and the anisotropy of conductivity was obtained in smectic A as well as nematic phase.

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1. Introduction

Since the liquid crystal displays play an important role in computer technology, new liquid crystalline materials showing high dielectric permittivity and anisotropy are required. For that reason, measurements of dielectric permittivity tensor principal components: $\varepsilon'_{\perp}(\omega)$ and $\varepsilon'_{\parallel}(\omega)$ as well as complex ionic conductivity σ are of the great importance [1–5]. Before a material can be selected for applications one should study its frequency, bias field, measuring field and temperature characteristics.

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Fig. 1. Molecular structure and phase sequence obtained for 5OSCl (a) and 6OSCl (b).

In scope of this paper, two compounds: 4'-chlorophenol-4-*n*-alkoxythiobenzoates (5OSCl and 6OSCl) having strongly polar bond (C–Cl) at *para* position [6] have been studied. Their molecular structures and phase sequences obtained on heating and cooling are presented in Fig. 1.

2. Experimental

Dielectric measurements were done by using Agilent 4294A impedance analyzer in the frequency range between 40 Hz and 15 MHz for two principal alignments of the samples. The measurements were performed during heating and cooling of the samples. Temperature was controlled by Eurotherm 2604, and measured by a thermocouple attached to Agilent 34401A multimeter controlled by a computer. The computer with adequate software that enables data acquisition, transferring, and displaying interesting material parameters such as dielectric permittivity, conductivity, loss factor, and other, controlled the experiment. The samples were put in three different types of the gold-coated 5 μ m cells, namely the planar cell (Au-HG) and homeotropic cell (Au-HT) and the (Au) cell with no surface treatment. Planar and homogeneous alignment was obtained by slowly cooling of the sample in Au-HG cells from the N to SmA phase. On the other hand, homeotropic alignment in nematic and smectic phase was obtained by applying a DC bias electric field of the order of 10⁶ V/m to the sample planar aligned in Au cells.

3. Results and discussion

Dielectric measurements were done for two principal alignments of substances studied to obtain parallel and perpendicular components of the electric permittivity. For homeotropic alignments a strong dielectric relaxation process was found in the SmA phase for both substances (Fig. 2). A Cole–Cole function with conductivity term (Eq. (1)) was fitted to the experimental data

$$\varepsilon^*(\omega) = \varepsilon(\infty) + \frac{\Delta\varepsilon_0}{1 + (\mathrm{i}\omega\tau_\mathrm{R})^{1-\alpha}} - \mathrm{i}\frac{\sigma(\omega)}{\varepsilon_0\omega},\tag{1}$$

where $\varepsilon(\infty)$ is a high frequency electric permittivity, ε_0 — electric permittivity of the free space, $\Delta \varepsilon_0$ — dielectric strength, $\tau_{\rm R}$ — relaxation time, α — distribution parameter of relaxation times ($0 \le \alpha \le 1$), $\sigma(\omega)$ — ionic conductivity, and

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Fig. 2. Dielectric dispersion and absorption versus frequency (ν) obtained for homeotropic alignment in SmA phase for 5OSCl (a) and 6OSCl (b).

Fitting parameters obtained for both substances at temperature of 70.0° C.

Substance	$\Delta \varepsilon_{0\parallel}$	$\varepsilon_{\infty\parallel}$	$\nu_{\rm R\parallel}~[{\rm MHz}]$	α	$\sigma_{\parallel} ~[{ m S/m}]$
5OSCl	5.63	3.26	0.910	0.012	3.21×10^{-9}
6OSCl	5.25	3.13	0.917	0.014	1.79×10^{-9}

 $\omega = 2\pi\nu$ is a circular frequency. The fitting parameters are presented in Table for one temperature. Solid lines in Fig. 2a and b are theoretical fits of Eq. (1).

The activation energy calculated for this process is close to 100 kJ/mol for both substances studied and is characteristic of the reorientation of molecule around the short molecular axis [7]. In both cases the α parameter is very small. This means that the relaxation process observed is a single Debye-type process.

Perpendicular and parallel components of the static electric permittivity are presented in Fig. 3. To obtain homeotropic alignment of the sample in Au-HG and Au cells the bias field was applied in the nematic phase. The homeotropic alignment of the SmA phase was obtained by slowly cooling of the aligned N phase. As seen for cells with planar alignment (Au-HG) the electric field applied was less effective than surface interaction in Au-HT cells to obtain the homeotropic state. It is worth noting that the parallel component of electric permittivity is distinctly higher than perpendicular one. Moreover, there is a step at the transition from the nematic to the smectic A phase and both components of electric permittivity have a lower value in the SmA phase. Such behavior shows that SmA phase is a SmA₂ one showing antiferroelectric order, because decrease in $\varepsilon_{\parallel}(T)$ with temperature lowering is connected with short range dipole–dipole negative correlation of the molecules [7–10]. As seen in Fig. 3 in the case of nematic phase, Au-HT cells give very good alignment. However, upon the N–SmA phase transition the homeotropic alignment is lost.

TABLE

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Fig. 3. Static electric permittivities versus temperature obtained for planar and homeotropic alignment for 5OSCl (a) and 6OSCl (b) in the N and SmA phases.

Dielectric anisotropy is defined as $\Delta \varepsilon = \varepsilon_{\parallel} - \varepsilon_{\perp}$. It is worth noting that the dielectric anisotropy is positive in nematic as well as in smectic A phase for both substances studied (Fig. 3). Due to the large and positive dielectric anisotropy, one can observe a nice Fréedericksz transition from planar to homeotropic texture [7]. Ionic conductivity was obtained by fitting Eq. (1) to the experimental data for planar and homeotropic configuration of the samples. Figure 4 illustrates temperature dependence of ionic conductivity components (σ_{\parallel} and σ_{\perp}). It is interesting that for 5OSCl the ionic conductivity along the director in nematic phase is an activation process, whereas for 6OSCl is non-activation one. Such behavior can result from different types of ions, which take part in conductivity process for 5OSCl than for 6OSCl.



Fig. 4. Temperature dependence of ionic conductivity obtained for planar and homeotropic alignment for 5OSCl (a) and 6OSCl (b).

One can also define conductivity anisotropy as

$$\Delta \sigma = \sigma_{\parallel} - \sigma_{\perp}, \tag{2}$$

where σ_{\parallel} and σ_{\perp} are conductivities for homeotropic and planar alignment of the sample, respectively.



Fig. 5. A ratio of parallel and perpendicular components of ionic conductivity versus temperature for 5OSCl.

From Figs. 4 and 5 one can conclude that for both substances studied the anisotropy of conductivity is positive in the nematic phase and negative in SmA phase. It means that in nematic phase the ionic conductivity is higher along the director, whereas in SmA phase the diffusion of ions is easier within the smectic layers in (x, y) planes of the SmA₂ phase. The ratio of conductivities $\sigma_{\parallel}/\sigma_{\perp} \approx 0.2$ is typical of biological membranes.

4. Conclusions

Summing up, the dielectric measurements done in the frequency range from 40 Hz to 15 MHz, allowed to determine the dielectric and conductivity anisotropy in liquid crystalline phases of strongly polar thioesters. For both substances, one molecular relaxation process was observed for homeotropic alignment, and it is interpreted as the reorientation of molecules around short molecular axis. Dielectric anisotropy is positive in both phases of two studied compounds. However, it decreases with decreasing temperature which is typical of the systems with strong dipole–dipole correlations. On the other hand, ionic conductivity ($\Delta \sigma$) changes the sign at the transition from SmA to N phase. In SmA phase $\Delta \sigma$ is negative, which means that the conductivity within smectic layers (planar alignment of molecules) is higher than crosswise the smectic layers (homeotropic configuration).

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