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Au–In System Studied with Perturbed Angular Correlation Method

P. WODNIECKI, A. KULIŃSKA AND B. WODNIECKA

The Henryk Niewodniczański Institute of Nuclear Physics Radzikowskiego 152, 31-342 Kraków, Poland

The Au–In system was the subject of our comprehensive study with the perturbed angular correlation technique. Bulk samples of all intermetallic phases shown in the phase diagram were prepared to determine the hyperfine interaction parameters and their temperature dependence. A polymorphic phase transformation in the AuIn phase at about 630 K and a new metastable modification of Au_7In_3 after quenching the sample of 30 at.% In from temperature above 700 K were found. In addition, formation of the Au–In phases of different stoichiometries at the interfaces of thin Au/In films was observed. It turned out that the starting composition of bilayer samples, i.e. the thickness ratio of the Au and In layers, strongly influenced the stoichiometry of the formed compounds.

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1. Introduction

The Au–In binary system exhibits a large number of intermetallic phases [1]. Unfortunately, in spite of many experimental data, some regions of its phase diagram are still not sufficiently examined. Some phases are reported to have different crystallographic structures (Au₇In), some are evidenced not in all phase diagrams (Au₃In₂), and the crystallographic structure of AuIn compound is still not exactly known. It was our aim to measure the electric field gradients (EFG) for ¹¹¹In probes in all Au–In compounds and to determine the number of nonequivalent In sites in the respective lattices. The perturbed angular correlation (PAC) method, due to its huge sensitivity for the changes of microstructure in the nearest neighborhood of the radioactive probes, is a perfect tool to study the polymorphic transformations, even in their very early stages. Thus, measurements in a wide temperature range allow us to observe the structural phase transitions and to determine the homogeneity and temperature region of the phase under study.

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Application of the PAC technique to detect the intermetallic compounds formed in the interface region after annealing at elevated temperatures also requires the knowledge of hyperfine interaction of the PAC probes, present in the lattices of the bulks. The electric field gradients at ¹¹¹In/¹¹¹Cd was measured for bulk $\operatorname{Au}_x \operatorname{In}_{(1-x)}$ samples of different stoichiometries [2]. The actual position of probe nuclei was obvious due to the fact that the parent ¹¹¹In is chemically identical with one of the compound constituents. The measurements were performed for each intermetallic compound shown in the Au–In phase diagram [1].

2. Experimental details

The Au–In bulk samples were prepared by melting under argon atmosphere the appropriate amounts of the constituents. The carrier-free solution of ¹¹¹In in 0.04 M HCl was earlier dried-out on the Au-foil in order to dope the investigated specimens with the ¹¹¹In-parent isotope of ¹¹¹Cd. An appropriate annealing in sealed and evacuated quartz tubes ensured the homogeneity of the samples. The PAC measurements were performed at different temperatures from 24 K to 700 K.

The Au/In bilayer films were deposited onto crystalline silicon wafers at room temperature. The relative thicknesses, measured by a quartz oscillator, were adjusted to obtain the desired composition of the samples. Gold layers, 50-80 nm thick, were covered by indium films with thicknesses of 38-154 nm equivalent to the desirable Au:In atomic ratio (1:2, 1:1, 2:1, 2.5:1, 3:1, 4:1). Some 10^{12} of 111 In atoms were deposited as a submonolayer at the interface via the evaporation from a thin tungsten wire. The PAC spectra were taken at room temperature before and after subsequent 10-min annealings.

3. Results and discussion

3.1. The bulk samples

The hexagonal crystallographic structure of the Au₇In phase is reported as DO_{24} (Ni₃Ti)-type [3] or the Nd-type [4]. The PAC spectrum (Fig. 1a) obtained for the Au_{88.1}In_{11.9} sample was described by a large frequency distribution around 11.3(6) MHz ($\delta = 33(3)\%$). Lack of the well-defined EFG excludes the ordered Ni₃Ti-type of structure, as two unique axially symmetric EFGs should be observed in that case. Thus, for this phase the Nd-type of structure proposed by Hiscocks and Hume-Rothery [4] is more likely.

The Au₃In phase is reported as orthorhombic DO_a structure [5]. This phase undergoes an order-disorder transformation between 572 K and 612 K [4]. In Au₇₅In₂₅ compound one nonaxial EFG was evidenced in the PAC pattern (Fig. 1b). The value of quadrupole frequency linearly decreases with the rise of temperature. An ordered-disordered phase transformation in Au₃In, manifested by the increase

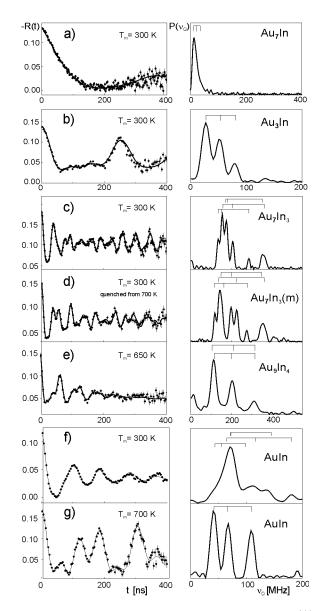


Fig. 1. PAC spectra with the corresponding Fourier transforms for ¹¹¹In doped Au–In bulk samples of different In concentrations taken at indicated temperatures.

in the width δ of quadrupole frequency distributions above 600 K, was observed in agreement with the phase diagram [1].

For the In concentration of ~ 30 at.% the low- and high-temperature (lt- and ht-) phase are reported. According to the phase diagram at ~ 650 K the ht-phase (Au₉In₄) undergoes a transformation to the lt-phase of Au₇In₃ stoichiometry hav-

ing a different structure. The PAC measurements performed on the Au₇₀In₃₀ sample below 650 K resulted in three nonaxial EFGs with equal contributions (Fig. 1c). Above 650 K two EFGs attributed to indium sites in the Au₉In₄ phase (Fig. 1e) were detected. The observed EFG distributions were large, as for the other γ -brasses [6]. The same sample quenched in water from the temperatures above 700 K exhibited three new EFGs (see Fig. 1d) which indicates the formation of a new phase. This phase is metastable and undergoes a transformation to the low-temperature phase above 350 K.

The crystallographic structure of the AuIn phase is reported to be triclinic [7], but its structure type is not known. The PAC measurements were performed at different temperatures between 300 K and 700 K. A change of PAC pattern observed at about 630 K exhibits a polymorphic phase transformation (Fig. 1f–g). Above this temperature one well-defined quadrupole frequency of 34.7(4) MHz with asymmetry parameter $\eta = 0.47(1)$ was measured. Below 630 K three quadrupole frequencies were necessary to fit the PAC spectra. The detailed PAC experiments proved that this transformation is reproducible and fully reversible. This finding disagrees with the Au–In phase diagram [1], where only a single AuIn phase in the whole temperature range of 300–783 K is reported. To confirm the existence of this phase transition the differential scanning calorimeter measurements were performed with three different scanning rates for heating and cooling the sample (Fig. 2). The temperature, in which the phase transition takes place, determined by this method (631.8(4) K) agrees very well with the value obtained via PAC.

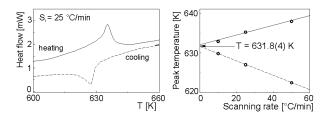


Fig. 2. DSC results obtained for AuIn compound; left: measurements performed for the scanning rate 25°C/min; right: results obtained for three different scanning rates used during heating and cooling process.

The results for Au₆₂In₃₈ stoichiometry did not confirm the existence of the ordered phase reported by some authors [7]. Also an attempt to observe the β_1 phase of 22 at.% indium concentration with Cu₁₀Sb₃ type of structure [8], where only unique nonaxially symmetric EFG is expected, failed.

In the set of experiments for Au/In bilayers having ¹¹¹In probes deposited at the interfaces the temperature-induced formation of different Au–In phases was evidenced [9]. A fast interdiffusion of the component metals caused bypassing of some phases and the stoichiometry of the "final" compound (i.e. after the final

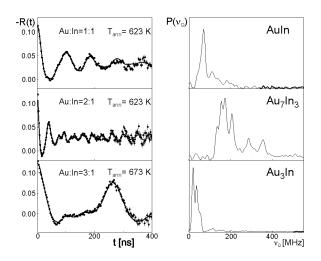


Fig. 3. Room temperature PAC spectra (with corresponding Fourier transforms) of ¹¹¹Cd in three Au/In bilayer samples taken after the final annealing at indicated temperatures.

annealing) formed in the bilayer, was governed by the starting Au:In atomic ratio. Figure 3 shows the PAC spectra for three Au/In bilayers of different atomic compositions, taken after formation of the corresponding stoichiometric phases.

The investigations presented here prove that the PAC method is very suitable for a detail study of the phase diagrams and of the phase formation at the interfaces of thin films.

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